

Review

# Time-resolved step-scan Fourier transform infrared investigation of heme-copper oxidases: implications for O<sub>2</sub> input and H<sub>2</sub>O/H<sup>+</sup> output channels

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Received 11 March 2003; received in revised form 21 May 2003; accepted 25 June 2003

## Abstract

We have applied FTIR and time-resolved step-scan Fourier transform infrared (TRS<sup>2</sup>-FTIR) spectroscopy to investigate the dynamics of the heme-Cu<sub>B</sub> binuclear center and the protein dynamics of mammalian *aa*<sub>3</sub>, *Pseudomonas stutzeri cbb*<sub>3</sub>, and *caa*<sub>3</sub> and *ba*<sub>3</sub> from *Thermus thermophilus* cytochrome oxidases. The implications of these results with respect to (1) the molecular motions that are general to the photodynamics of the binuclear center in heme-copper oxidases, and (2) the proton pathways located in the ring A propionate of heme *a*<sub>3</sub>-Asp372-H<sub>2</sub>O site that is conserved among all structurally known oxidases are discussed.

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## 1. Introduction

Cytochrome oxidase sustains mitochondrial electron transport and linked ATP synthesis by catalysing the four-electron reduction of dioxygen to water [1]. Among oxygen-metabolizing heme enzymes, cytochrome oxidase is unique in being susceptible to high time resolution spectroscopic investigation of its reaction time course and of the intermediates, partially reduced oxygen structures that occur in the binuclear center during oxygen reduction [2–10]. Substantial progress has been made in gaining a molecular level of understanding of the structures for many intermediates in the O<sub>2</sub>/enzyme reaction. The time-resolved resonance Raman (TR<sup>3</sup>) approach used by Babcock's, Rousseau's, and Kitagawa's groups, although under different experimental conditions, allowed the identification of the heme *a*<sub>3</sub> iron-bound oxy, ferryl, and hydroxyl species that occur at various stages in the mechanism [2–10]. There appears now to be general consensus as to the occurrence of these species, although the precise details of the structure of the

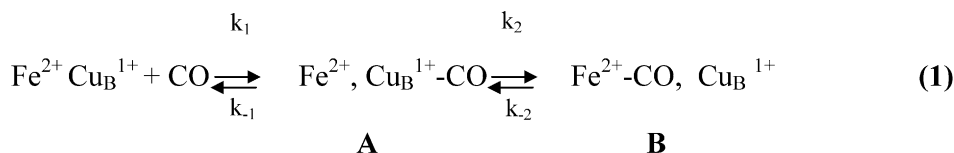
P intermediate and the relationship between the electron transfer events and its formation remain to be determined unambiguously.

The binding and release of small molecular ligands from the binuclear heme-Cu<sub>B</sub> center in cytochrome oxidase is a complex process that involves ligand motion through the protein and extensive changes in the heme electronic structure and molecular conformation. Studies of CO adducts of cytochrome oxidase have been shown to be important probes of the active site, and photodissociation and recombination studies have revealed the dynamic processes that occur in the binuclear center [11–22]. In order to determine the molecular basis for the function of heme-copper oxidases, it is necessary to understand the interactions between the heme-bound ligands and the residues in their distal environment. Ultimately, this should lead to an understanding of the ligand kinetic behavior and the active site(s) properties. The binding and photodynamics of CO to heme-copper oxidases proceed according to Schemes 1 and 2, respectively. Where *k*<sub>1</sub> and *k*<sub>−1</sub> represent the reversible binding of CO to Cu<sub>B</sub>, and *k*<sub>2</sub> is the first-order transfer of CO from Cu<sub>B</sub> to the heme Fe [23]. In Scheme 2, the Cu<sub>B</sub><sup>1+</sup>-CO complex is fully developed within 1 ps subsequent to CO photolysis from heme *a*<sub>3</sub>, demonstrating the absence of activation barriers to the CO transfer from heme *a*<sub>3</sub> to Cu<sub>B</sub> [23]. The state denoted by an asterisk represents a non-

Abbreviations: CcO, cytochrome *c* oxidase; FTIR, Fourier transform infrared; TR-FTIR, time-resolved Fourier transform infrared; MCT, mercury cadmium telluride; RR, resonance Raman

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Scheme 1.

equilibrium heme  $a_3$  state characterized by an upshifted Fe–His stretching vibration which, in the mammalian  $aa_3$  oxidase, relaxes to the equilibrium reduced species at times  $>10 \mu\text{s}$  [24].

Recently, we and others have applied time-resolved step-scan Fourier transform infrared spectroscopy (TRS<sup>2</sup>-FTIR) to investigate the protein dynamics of heme-copper oxidases at room temperature [14,17,19–22]. It has been demonstrated that the dynamics of a large protein complex such as cytochrome oxidase can be resolved on the single vibrational level with TRS<sup>2</sup>-FTIR spectroscopy. In this work, we present a comprehensive TRS<sup>2</sup>-FTIR study of bovine cytochrome  $aa_3$ , cytochrome  $cbb_3$  from *Pseudomonas stutzeri*, and  $ba_3$  and  $caa_3$  from *Thermus thermophilus*. The implications of these results with respect to the input  $\text{O}_2$  and output  $\text{H}^+/\text{H}_2\text{O}$  channels are discussed.

## 2. Materials and methods

Mammalian CcO was isolated from beef hearts [7]. The purification of  $cbb_3$ ,  $ba_3$ , and  $caa_3$  oxidase was according to Refs. [19–21]. The pD solutions prepared in  $\text{D}_2\text{O}$  buffers were measured by using a pH meter and assuming  $\text{pD} = \text{pH}$  (observed) + 0.4. Dithionite reduced samples were exposed to 1 atm CO (1 mM) in an anaerobic cell to prepare the carbonmonoxy adduct and loaded anaerobically into a cell with  $\text{CaF}_2$  windows and a 0.025-mm spacer. CO gas was obtained from Messer (Germany) and isotopic CO ( $^{13}\text{CO}$ ) was purchased from Isotec. FTIR spectra were obtained from 400–500  $\mu\text{M}$  samples with a Bruker Equinox 55 FTIR spectrometer equipped with a liquid nitrogen cooled mercury cadmium telluride detector. The 532-nm pulse from a Continuum Nd-YAG laser (7-ns width, 3 Hz) was used as a pump light (3–4 mJ/pulse) to photolyze the heme  $a_3$ -CO complex. The time-resolved FTIR spectra were obtained with spectral resolution of  $8 \text{ cm}^{-1}$  and 5- $\mu\text{s}$  time resolution for the 5  $\mu\text{s}$ –4 ms measurements or 100  $\mu\text{s}$  for the 100  $\mu\text{s}$ –80 ms measurements. Total number of slices was 800, with 40 slices before and 760 after the laser firing. A total of 10 co-additions per retardation data point was collected. Changes in intensity were recorded with an MCT detector, amplified (dc-coupled) and digitized with a 200-kHz, 16-bit,

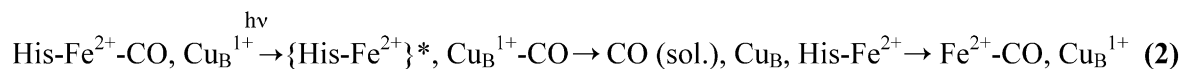
analog-to-digital converter. Blackman–Harris three-term apodization function with  $32\text{-cm}^{-1}$  phase resolution and the Mertz phase correction algorithm were used. Difference spectra were calculated as  $\Delta A = -\log(I_S/I_R)$ . Optical absorbance spectra were recorded before and after FTIR measurements in order to assess sample stability with a Perkin-Elmer Lambda 20 UV–VIS spectrometer.

## 3. Results and discussion

Fourier transform infrared difference spectroscopy (FTIR) is a powerful structure-specific technique for exploring changes that occur to individual amino acid residues in a protein as a result of changes to redox and ligation states [13–22]. The FTIR difference approach has also been used to investigate the CO-photoproduct and the electrochemical oxidized-minus-reduced difference spectra of heme-copper oxidases. In the latter case, the perturbation is the redox state of the metal centers, whereas in the former is the photodissociation of CO from the heme. With the time-resolved step-scan approach, recently it was demonstrated that although the exogenous ligand vibrations (CO) were essentially identical between the room- and low-temperature spectra of photodissociated CO-cytochromes  $aa_3$  and  $bo_3$ , significant differences exist in the protein bands between these temperatures. It was suggested that these differences originate from the fact that at room temperature, CO has dissociated from  $\text{Cu}_B$ , whereas in the low-temperature (80 K) the final state has CO coordinated to  $\text{Cu}_B$  [14].

### 3.1. Mammalian cytochrome $aa_3$ dynamics

Infrared and UV–VIS experiments of CO-bound  $aa_3$  have shown that CO dissociates from  $\text{Cu}_B$  on a microsecond time scale ( $k_{-1} = 5 \times 10^5 \text{ s}^{-1}$ ), and a thermal equilibrium between  $\text{Cu}_B$ -bound and “free” CO in solution ( $K = 87 \text{ M}^{-1}$ ) is established [23]. The CO, however, does not return to heme  $a_3$  for milliseconds ( $k_2 = 1030 \text{ s}^{-1}$ ). Fig. 1 shows the TRS<sup>2</sup> spectra ( $t_d = 5 \mu\text{s}$ –4 ms,  $8 \text{ cm}^{-1}$  spectral resolution) of mammalian fully reduced  $aa_3$ -CO subsequent to CO photolysis in  $\text{D}_2\text{O}$ . We detect no intensity changes of the  $1965 \text{ cm}^{-1}$  peak from  $t_d = 5$ –750  $\mu\text{s}$ , which implies the lack



Scheme 2.

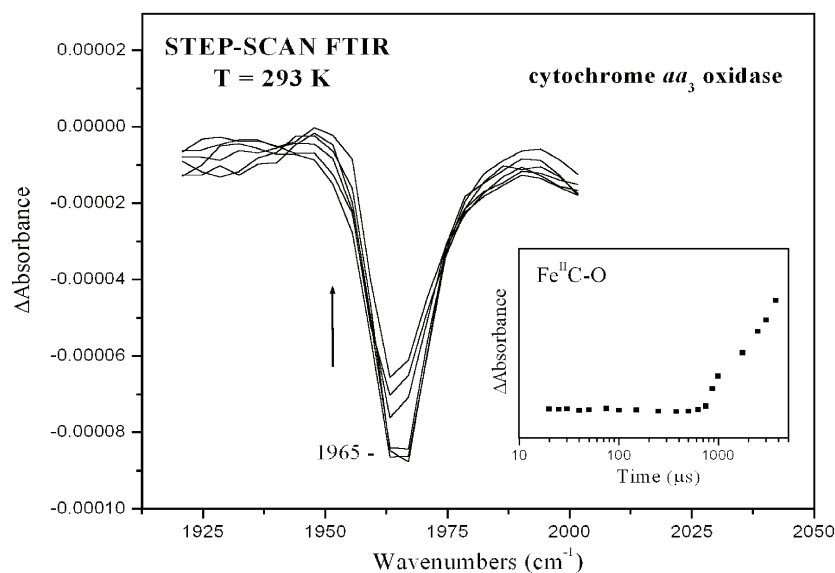


Fig. 1. Time-resolved step-scan FTIR difference spectra of the photolyzed CcO-CO complex at 25, 100, 500, 1000, 2000, and 3500  $\mu\text{s}$ . Inset: Plot of the 1965  $\text{cm}^{-1}$  ( $\nu_{\text{Fe-C-O}}$ , squares) mode vs. time subsequent to CO photolysis.

of CO rebinding to heme  $a_3$ . At times longer than 750  $\mu\text{s}$ , the intensity increase of the 1965- $\text{cm}^{-1}$  peak signals the onset of ligand rebinding to heme  $a_3$ . This behavior is analogous to that observed by Bailey et al. [14] and Findsen et al. [24]. The former authors proposed that on the time scale CO leaves  $\text{Cu}_\text{B}$  ( $t_{1/2} = 1.5 \mu\text{s}$ ), there are no intrinsic heme barriers to recombination with heme  $a_3$  because typical geminate recombination rates are nanoseconds. Thus, another barrier must form which prevents CO recombination prior to 1 ms, and suggested as possible candidates either a local protein conformational change or the binding of an endogenous ligand which blocks the distal ligation site. Time-resolved ps RR experiments [25] have demonstrated that heme  $a_3^{2+}$  is five-coordinate, high spin, with proximal histidine ligation to the heme at the earliest times following CO photodissociation, and thus, no endogenous ligand blocks the distal binding site of heme  $a_3$ . Thus, it is highly unlikely that the barrier is the result of an endogenous ligand binding that blocks the distal ligation site, and a local protein conformational change is the most probable cause of its formation. Obviously, the heme  $a_3$ - $\text{Cu}_\text{B}$  pocket is constructed in such a manner that results in rapid migration of the photodissociated CO from heme  $a_3$  to  $\text{Cu}_\text{B}$  and slow recombination of CO to heme  $a_3$ .

### 3.2. Cytochrome *cbb*<sub>3</sub> dynamics

The microaerobic *cbb*<sub>3</sub>-type cytochrome *c* oxidase isolated from *P. stutzeri* is an isoenzyme in the family of cytochrome *c* oxidases (CcO) [19]. It contains three *c*-type low-spin hemes, one low-spin *b*-type heme, and a heme  $b_3$ - $\text{Cu}_\text{B}$  binuclear center. The structural characteristics of the binuclear center in *cbb*<sub>3</sub> are unique when compared to those of *aa*<sub>3</sub> type oxidases; it lacks the hydroxyethylfarnesyl side chain

and the highly conserved among the heme-copper oxidases tyrosine 244, both of which have been proposed to play a crucial role in the properties of the binuclear center. Data on the photodynamics of the CO-bound adduct of cytochrome *cbb*<sub>3</sub> in conjunction with those of the *aa*<sub>3</sub>-type oxidase can be interpreted to yield specific information concerning electronic and heme/ $\text{Cu}_\text{B}$  geometric properties, and heme/ $\text{Cu}_\text{B}$ -axial ligand bonding interactions. The time-resolved FTIR data revealed that in *cbb*<sub>3</sub>-type oxidase the decay of the transient  $\text{Cu}_\text{B}$ -CO complex is concurrent with the formation of the heme  $b_3$ -CO complex, and the  $\nu(\text{CO})$  of  $\text{Cu}_\text{B}$  at 2065  $\text{cm}^{-1}$ , despite the lack of the cross-link tyrosine 244 is similar to that observed in cytochrome *aa*<sub>3</sub> [23] and cytochrome *bo*<sub>3</sub> [14]. The former observation contrasts sharply with the well-known behaviour of CO recombination to the heme Fe of cytochrome *aa*<sub>3</sub> and *bo*<sub>3</sub>-type oxidases. The *cbb*<sub>3</sub> is the first heme-copper oxidase reported in which the decay of the  $\text{Cu}_\text{B}$ -CO complex is accompanied by the formation of the Fe-CO complex at room temperature. It is also the only oxidase with an open heme-copper structure lacking the  $\alpha$ -form of the heme, thus the distal interactions due to  $\text{Cu}_\text{B}$  and its local environment, and also lacking the hydrogen bonding between the farnesyl hydroxyl and the tyrosine hydroxyl which couples together the heme  $a_3$  and  $\text{Cu}_\text{B}$ .

### 3.3. Cytochrome *caa*<sub>3</sub> dynamics

Fourier transform infrared and step-scan time-resolved FTIR difference spectra have been reported for the carbon-monooxy cytochrome *caa*<sub>3</sub> from *T. thermophilus* [21]. A major C-O mode of heme  $a_3$  at 1958  $\text{cm}^{-1}$  and two minor modes at 1967 and 1975  $\text{cm}^{-1}$  (7:1:1) have been identified at room temperature, and remained unchanged in  $\text{H}_2\text{O}/\text{D}_2\text{O}$  exchange. The time-resolved FTIR data indicate that the

Table 1

Kinetic parameters for CO binding to mammalian *aa<sub>3</sub>*, *cbb<sub>3</sub>*, *caa<sub>3</sub>* and *ba<sub>3</sub>* from *T. thermophilus*

|                        | $k_2$ (s <sup>-1</sup> ) | $k_{-1}$ (s <sup>-1</sup> ) | Reference |
|------------------------|--------------------------|-----------------------------|-----------|
| <i>aa<sub>3</sub></i>  | 1030                     | $7 \times 10^5$             | [23]      |
| <i>cbb<sub>3</sub></i> | $4.8 \times 10^3$        | $5.8 \times 10^3$           | [19]      |
| <i>caa<sub>3</sub></i> | 34.1                     | $2.3 \times 10^4$ (35%)     | [21]      |
|                        |                          | 36.3 (65%)                  |           |
| <i>ba<sub>3</sub></i>  | 28.6                     | 34.5                        | [20]      |

transient  $\text{Cu}_B^{1+}$ -CO complex is formed at room temperature as revealed by the CO stretching mode at 2062 cm<sup>-1</sup>. Therefore, the *caa<sub>3</sub>* enzyme is the only documented member of the heme-copper superfamily whose binuclear center consists of an *a<sub>3</sub>*-type heme of a  $\beta$ -form and a  $\text{Cu}_B$  atom of an  $\alpha$ -form. These results illustrate that the properties of the binuclear center in other oxidases resulting in the  $\alpha$ -form are not required for enzymatic activity. Dissociation of the transient  $\text{Cu}_B^{1+}$ -CO complex is biphasic. The rate of decay is  $2.3 \times 10^4$  s<sup>-1</sup> (fast phase, 35%) and 36.3 s<sup>-1</sup> (slow phase, 65%). The observed rate of rebinding to heme *a<sub>3</sub>* is 34.1 s<sup>-1</sup>. Table 1 summarizes the CO kinetic properties in heme-copper oxidases.

### 3.4. Cytochrome *ba<sub>3</sub>* dynamics

We have reported the first evidence for the existence of the equilibrium  $\text{Cu}_B^{1+}$ -CO species of CO-bound reduced

cytochrome *ba<sub>3</sub>* from *T. thermophilus* at room temperature [20]. The frequency of the C–O stretching mode of  $\text{Cu}_B^{1+}$ -CO is located at 2053 cm<sup>-1</sup> and remains unchanged in H<sub>2</sub>O/D<sub>2</sub>O exchanges and between pH/pD 5.5 and 9.7, indicating that the chemical environment does not alter the protonation state of the  $\text{Cu}_B$  histidine ligands. The time-resolved step-scan FTIR difference spectra indicate that the rate of decay of the transient  $\text{Cu}_B^{1+}$ -CO complex is 34.5 s<sup>-1</sup> and rebinding to heme *a<sub>3</sub>* occurs with  $k_2$ =28.6 s<sup>-1</sup>. The rate of decay of the transient  $\text{Cu}_B^{1+}$ -CO complex displays similar time constant as the absorption changes at 1694(+)/1706(–), attributed to perturbation of the heme *a<sub>3</sub>* propionates (COOH). The  $\nu(\text{C–O})$  of the transient  $\text{Cu}_B^{1+}$ -CO species is the same as that of the equilibrium  $\text{Cu}_B^{1+}$ -CO species and remains unchanged in the pD range 5.5–9.7, indicating that no structural change takes place at  $\text{Cu}_B$  between these states. The transient binding of CO to  $\text{Cu}_B$  is dynamically linked to structural changes at the ring A propionate of heme *a<sub>3</sub>* (1694/1706 cm<sup>-1</sup>) and concomitantly to a change in the local environment of Asp372 (1726 cm<sup>-1</sup>).

We have also reported the first study of O<sub>2</sub> migration in the putative O<sub>2</sub> channel of cytochrome *ba<sub>3</sub>* [26,27], and its effect to the properties of the binuclear heme *a<sub>3</sub>*- $\text{Cu}_B$  center of cytochrome *ba<sub>3</sub>* from *T. thermophilus* [22]. The FTIR spectra of the *ba<sub>3</sub>*-CO complex demonstrate that in the presence of 60–80  $\mu\text{M}$  O<sub>2</sub>, the  $\nu(\text{C–O})$  of  $\text{Cu}_B^{1+}$ -CO at 2053 cm<sup>-1</sup> (Complex A) shifts to 2045 cm<sup>-1</sup>, and remains

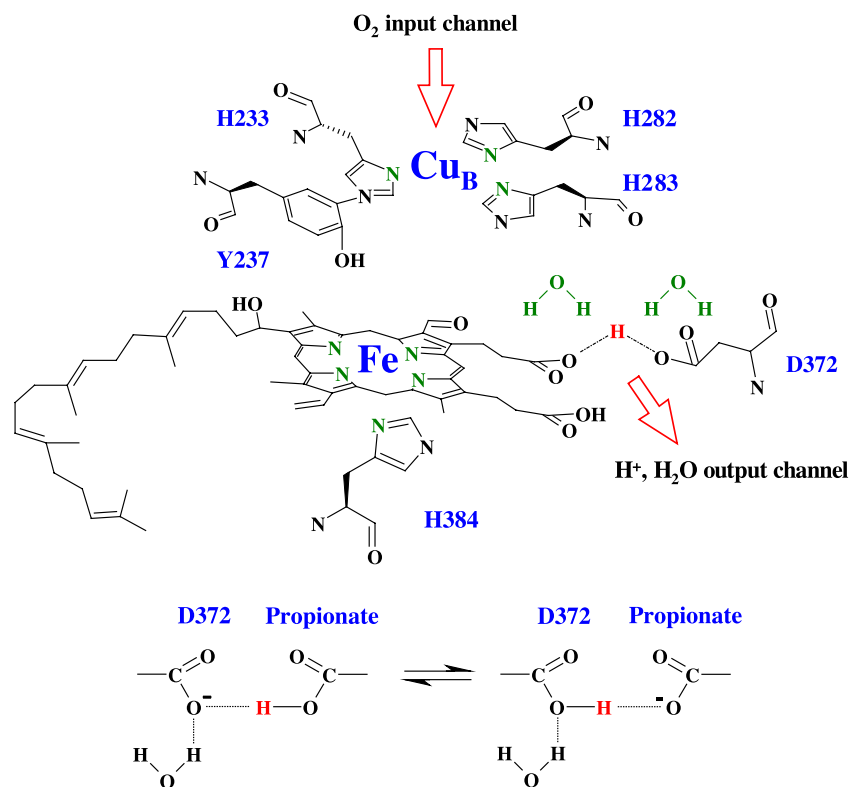


Fig. 2. Model for input/output channels in heme-copper oxidases. The residues are numbered following the sequence of cytochrome *ba<sub>3</sub>* oxidase from *T. thermophilus* (For color see online version).



unchanged in H<sub>2</sub>O/D<sub>2</sub>O exchanges and in the pH 6.5–9.0 range. The frequencies, but not the intensities, of the C–O stretching modes of heme *a*<sub>3</sub>–CO (complex **B**), however, remain unchanged. The change in the  $\nu(\text{CO})$  of complex **A** results in an increase of  $k_{-2}$ , and thus to a higher affinity of Cu<sub>B</sub> for exogenous ligands. The step-scan time-resolved FTIR difference spectra indicate that the rate of decay of the transient Cu<sub>B</sub><sup>1+</sup>–CO complex at pH 6.5 is 30.4 s<sup>−1</sup>, and 28.3 s<sup>−1</sup> in the presence of O<sub>2</sub>. The rebinding to heme *a*<sub>3</sub> occurs with  $k_2 = 26.3$  s<sup>−1</sup>, and 24.6 s<sup>−1</sup> in the presence of O<sub>2</sub>. The results provide solid evidence that in cytochrome *ba*<sub>3</sub> the ligand delivery channel is located at the Cu<sub>B</sub> site, which is the ligand entry to the heme *a*<sub>3</sub> pocket. We suggest that the properties of the O<sub>2</sub> channel are not limited to facilitate ligand diffusion to the active site, but are extended in controlling the dynamics and reactivity of the reactions of *ba*<sub>3</sub> with O<sub>2</sub> and NO.

### 3.5. Model for input/output channels in heme-copper oxidases

The determination of the structures in the binuclear center upon CO binding and the photodissociation/rebinding process remains an important problem that is central to elucidating the dynamics of heme-copper oxidases. The data presented here allow analysis of the general structural perturbations that are responsible for the unique conformations of heme-copper oxidases. Based on the existing data, we present a model (Fig. 2) for the O<sub>2</sub> input channel located at the Cu<sub>B</sub> site and an H<sup>+</sup>/H<sub>2</sub>O output channel located at the ring A propionate of heme *a*<sub>3</sub>–Asp372–H<sub>2</sub>O site (*T. thermophilus* numbering), which is conserved among all structurally known oxidases. TRS<sup>2</sup>-FTIR experiments are in progress to address these issues.

### Acknowledgements

This work is dedicated to the memory of Jerry. C.V. was a graduate student, post-doc, and collaborator of Jerry, but most importantly part of his extended family. I was touched by Jerry's friendship and inspired by his expectations of high standards. This work was partially supported from the Greek Ministry of Education.

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